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Key indicators

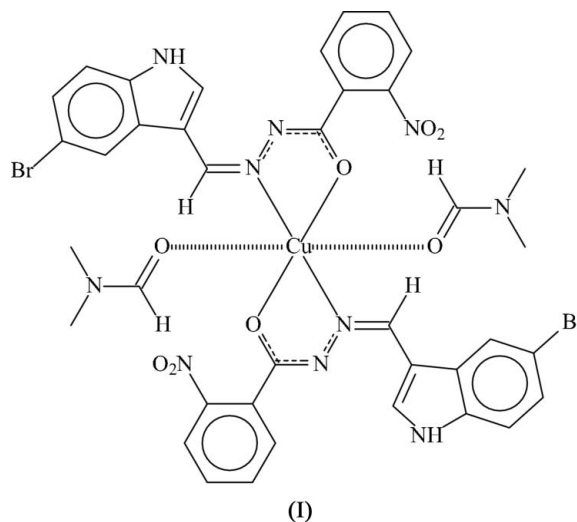
Single-crystal X-ray study
 $T = 295$ K
Mean $\sigma(\text{C}-\text{C}) = 0.006$ Å
 R factor = 0.055
 wR factor = 0.180
Data-to-parameter ratio = 13.9For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.Bis(5-bromo-1*H*-indole-3-carbaldehyde
2-nitrobenzoylhydrazonato- $\kappa^2\text{N},\text{O}$)bis-
(*N,N*-dimethylformamide- κO)copper(II)

The Cu atom in the title compound, $[\text{Cu}(\text{C}_{16}\text{H}_{10}\text{BrN}_4\text{O}_3)_2(\text{C}_3\text{H}_7\text{NO})_2]$, lies on a special position of $\bar{1}$ site symmetry in a grossly elongated CuN_2O_4 octahedral geometry [$\text{Cu}\cdots\text{O}_{\text{DMF}} = 3.014(4)$ Å]. The Cu atom is also *N,O*-chelated by the hydrazonate ligand. An $\text{N}-\text{H}\cdots\text{O}$ hydrogen bond helps to consolidate the crystal packing.

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Comment

A previous report (Ali *et al.*, 2005*a*) described the crystal structure of the Schiff base that is synthesized by condensing 5-bromoindole-3-carbaldehyde with 2-nitrobenzoylhydrazine. In the title compound, (I) (Fig. 1), two of these deprotonated Schiff bases chelate to copper (site symmetry $\bar{1}$) in a square-planar geometry (Table 1). However, the O atom of the DMF molecule lies at a distance of 3.014 (4) Å from the Cu atom; thus, the copper coordination geometry can also be regarded as grossly distorted octahedral. The molecules of (I) are linked by an $\text{N}-\text{H}\cdots\text{O}_{\text{DMF}}$ hydrogen bond (Table 2) into a chain.



The nickel derivative of 5-bromoindole-3-carbaldehyde benzoylhydrazone was recrystallized from pyridine to afford the bis-pyridine adduct; the geometry of the metal atom coordination is an essentially regular octahedron, and the pyridine N atoms are *cis* to each other (Ali *et al.*, 2005*b*).

Experimental

5-Bromoindole-3-carboxaldehyde was condensed with 2-nitrobenzoylhydrazide to form the Schiff base 5-bromo-1*H*-indole-3-carbaldehyde 2-nitrobenzoylhydrazone (Ali *et al.*, 2005*b*). This

reactant (0.40 g, 1.0 mmol) and copper(II) acetate dihydrate (0.11 g, 0.5 mmol) were heated in ethanol (50 ml) for several hours. The solid that separated from solution was purified by recrystallization from DMF. Green crystals of (I) were isolated after two weeks.

Crystal data

$[\text{Cu}(\text{C}_{16}\text{H}_{10}\text{BrN}_4\text{O}_3)_2(\text{C}_3\text{H}_7\text{NO})_2]$	$Z = 1$
$M_r = 982.11$	$D_x = 1.573 \text{ Mg m}^{-3}$
Triclinic, $P\bar{1}$	Mo $K\alpha$ radiation
$a = 8.643 (5) \text{ \AA}$	Cell parameters from 3209 reflections
$b = 9.102 (5) \text{ \AA}$	$\theta = 2.4\text{--}28.0^\circ$
$c = 14.302 (9) \text{ \AA}$	$\mu = 2.52 \text{ mm}^{-1}$
$\alpha = 99.84 (2)^\circ$	$T = 295 (2) \text{ K}$
$\beta = 99.27 (1)^\circ$	Block, green
$\gamma = 106.43 (1)^\circ$	$0.62 \times 0.44 \times 0.15 \text{ mm}$
$V = 1037 (1) \text{ \AA}^3$	

Data collection

Rigaku Mercury CCD diffractometer	3797 independent reflections
ω scans	2915 reflections with $I > 2\sigma(I)$
Absorption correction: multi-scan (CRYSTALCLEAR; Rigaku/MS, 2005)	$R_{\text{int}} = 0.050$
$T_{\text{min}} = 0.114$, $T_{\text{max}} = 0.686$	$\theta_{\text{max}} = 25.5^\circ$
17658 measured reflections	$h = -10 \rightarrow 10$
	$k = -10 \rightarrow 11$
	$l = -17 \rightarrow 15$

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_o^2) + (0.1064P)^2 + 0.1574P]$
$R[F^2 > 2\sigma(F^2)] = 0.055$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.180$	$(\Delta/\sigma)_{\text{max}} = 0.001$
$S = 1.10$	$\Delta\rho_{\text{max}} = 0.78 \text{ e \AA}^{-3}$
3797 reflections	$\Delta\rho_{\text{min}} = -0.64 \text{ e \AA}^{-3}$
274 parameters	
H atoms treated by a mixture of independent and constrained refinement	

Table 1

Selected geometric parameters (\AA , $^\circ$).

Cu1—O1	1.930 (3)	Cu1—N3	1.937 (3)
Cu1—O4	3.014 (4)		
O1—Cu1—O4	85.0 (1)	N3—Cu1—O4	88.4 (1)
O1—Cu1—N3	81.8 (1)	N3 ⁱ —Cu1—O4	91.7 (1)
O1—Cu1—N3 ⁱ	98.2 (1)		

Symmetry code: (i) $-x + 1, -y + 1, -z + 1$.

Table 2

Hydrogen-bond geometry (\AA , $^\circ$).

$D\text{—}H\cdots A$	$D\text{—}H$	$H\cdots A$	$D\cdots A$	$D\text{—}H\cdots A$
$\text{N4—H4n}\cdots\text{O4}^{\text{ii}}$	0.85 (1)	1.95 (2)	2.777 (4)	163 (5)

Symmetry code: (ii) $x, y + 1, z$.

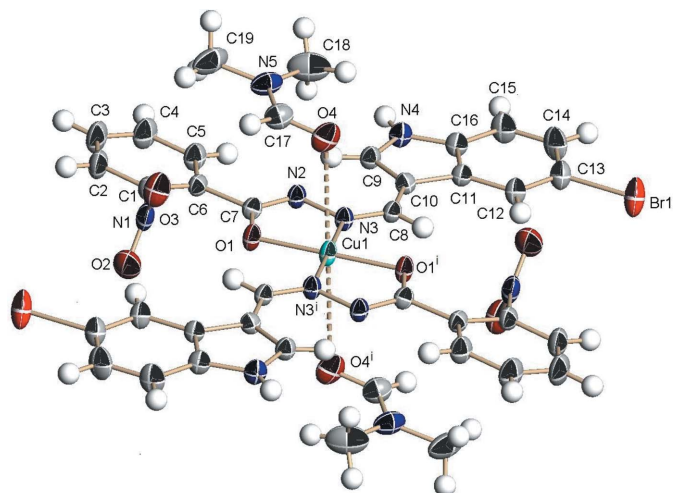


Figure 1

View of (I), showing 30% displacement ellipsoids (arbitrary spheres for the H atoms). The long $\text{Cu}\cdots\text{O}_{\text{DMF}}$ interaction is indicated by a dashed line [symmetry code: (i) $1 - x, 1 - y, 1 - x$].

The carbon-bound H atoms were positioned geometrically ($\text{C—H} = 0.93$ or 0.96 \AA) and refined as riding, with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ or $1.5U_{\text{eq}}(\text{methyl})$. The methyl groups were rotated to fit the electron density. The nitrogen-bound H atom was located in a difference Fourier map and refined with a distance restraint of $\text{N—H} = 0.85 (1) \text{ \AA}$; the U_{iso} value was freely refined.

Data collection: CRYSTALCLEAR (Rigaku/MS, 2005); cell refinement: CRYSTALCLEAR; data reduction: CRYSTALCLEAR; program(s) used to solve structure: CrystalStructure (Rigaku/MS, 2005); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: ORTEPII (Johnson, 1976); software used to prepare material for publication: SHELXL97.

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